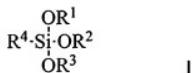


This listing of claims will replace all prior versions, and listings, of claims in the application:

Listing of Claims:

1. (Previously amended) A method of preparing siliceous materials comprising combining an organic polyol silane precursor with one or more additives under conditions suitable for hydrolysis and condensation of the precursor to a siliceous material, wherein the one or more additives are selected from one or more water-soluble polymers and one or more trifunctional silanes of Formula I:



wherein OR¹, OR² and OR³ are the same or different and represent a group that is hydrolyzed under normal sol-gel conditions to provide Si-OH groups; and R⁴ is group that is not hydrolyzed under normal sol-gel conditions, wherein the conditions suitable for hydrolysis and condensation of the precursor to a siliceous material comprise combining the organic polyol silane precursor with the one or more additives at a pH in the range of about 4 to about 11.5.

2. (Original) The method according to claim 1, wherein the one or more additives are water soluble polymers selected from one or more of polyethers, polyalcohols, polysaccharides, poly(vinyl pyridine), polyacids, polyacrylamides and polyallylamine.

3. (Original) The method according to claim 2, wherein the one or more additives are water soluble polymers selected from one or more of polyethylene oxide (PEO), polyethylene glycol (PEG), amino-terminated polyethylene oxide (PEO-NH₂), amino-terminated polyethylene glycol (PEG-NH₂), polypropylene glycol (PPG), polypropylene oxide (PPO), polypropylene glycol bis(2-amino-propyl ether) (PPG-NH₂), polyvinyl

alcohol, poly(acrylic acid), poly(vinyl pyridine), poly(N-isopropylacrylamide) (polyNIPAM) and polyallylamine (PAM).

4. (Original) The method according to claim 3, wherein the one or more additives are water soluble polymers selected from one or more of PEO, PEO-NH₂, PEG, PPG-NH₂, polyNIPAM and PAM.

5. (Original) The method according to claim 3, wherein the one or more additives are water soluble polymers selected from one or more of PEO, PEO-NH₂ and polyNIPAM.

6. (Original) The method according to claim 1, wherein the one or more additives is a mixture of water soluble polymers,

7. (Original) The method according to claim 6 wherein the mixture of water soluble polymers comprises PEO and PEO-NH₂.

8. (Original) The method according to claim 5, wherein the one or more additives is PEO.

9. (Original) The method according to claim 8, wherein the PEO has a molecular weight that is greater than about 10,000 g/mol.

10. (Original) The method according to claim 9, wherein the PEO is used at a concentration of greater than about 0.005 g/mL of final solution.

11. (Original) The method according to claim 5, wherein the one or more additives is PEO-NH₂.

12. (Original) The method according to claim 11, wherein the PEO-NH₂ has a molecular weight that is greater than about 3,000 g/mol and is used at a concentration of about 0.005 g/mL of final solution.

13. (Original) The method according to claim 5, wherein the one or more additives is poly(N-isopropylacrylamide).

14. (Original) The method according to claim 13, wherein the poly(N-isopropylacrylamide) has a molecular weight that is about 10,000 g/mol and is used at a concentration of about 0.005 g/mL of final solution.

15. (Original) The method according to claim 1, wherein the one or more additives is a compound of Formula I.

16. (Original) The method according to claim 15, wherein OR¹, OR² and OR³ are the same or different and are derived from organic di- or polyols.

17. (Original) The method according to claim 16, wherein OR¹, OR² and OR³ are the same or different and are derived from sugar alcohols, sugar acids, saccharides, oligosaccharides or polysaccharides.

18. (Previously amended) The method according to claim 16, wherein OR¹, OR² and OR³ are the same or different and are derived from allose, altrose, glucose, mannose, gulose, idose, galactose, talose, ribose, arabinose, xylose, lyxose, threose, erythrose, glyceraldehydes, sorbose, fructose, dextrose, levulose, sorbitol, sucrose, maltose, cellobiose, lactose, dextran (500-50,000 MW), amylose, pectin, glycerol, propylene glycol or trimethylene glycol.

19. (Original) The method according to claim 18, wherein OR¹, OR² and OR³ are the same or different and are derived from glycerol, sorbitol, maltose, trehalose, glucose, sucrose, amylose, pectin, lactose, fructose, dextrose and dextran.

20. (Original) The method according to claim 18, wherein OR¹, OR² and OR³ are the same or different and are derived from glycerol, sorbitol, maltose or dextran.

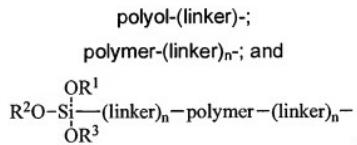
21. (Original) The method according to claim 15, wherein OR¹, OR² and OR³ are the same or different and are selected from C₁₋₄alkoxy, aryloxy and arylalkyleneoxy.

22. (Original) The method according to claim 21, wherein OR¹, OR² and OR³ are the same or different and are selected from C₁₋₄alkoxy, phenoxy, naphthoxy and benzyloxy.

23. (Original) The method according to claim 22, wherein OR¹, OR² and OR³ are the same or different and are selected from C₁₋₄alkoxy.

24. (Original) The method according to claim 23, wherein OR¹, OR² and OR³ are all ethoxy.

25. (Original) The method according to claim 15, wherein R⁴ is selected from the group consisting of:



wherein n is 0-1.

26. (Original) The method according to claim 25, wherein the polyol is an organic di- or polyol.

27. (Original) The method according to claim 26, wherein the polyol is selected from the group consisting of a sugar alcohol, sugar acid, saccharide, oligosaccharide and polysaccharide.
28. (Original) The method according to claim 27, wherein the polyol is selected from the group consisting of allose, altrose, glucose, mannose, gulose, idose, galactose, talose, ribose, arabinose, xylose, lyxose, threose, erythrose, glyceraldehydes, sorbose, fructose, dextrose, levulose, sorbitol, sucrose, maltose, cellobiose, lactose, dextran, (500-50,000 MW), amylose, pectin, glycerol, propylene glycol and trimethylene glycol.
29. (Original) The method according to claim 28, wherein the polyol is selected from the group consisting of glycerol, sorbitol, maltose, trehalose, glucose, sucrose, amylose, pectin, lactose, fructose, dextrose and dextran.
30. (Previously amended) The method according to claim 29, wherein the polyol is selected from the group consisting of glycerol, sorbitol, glucose, maltose and dextrose.
31. (Original) The method according to claim 25 wherein the polymer is a water soluble polymer.
32. (Original) The method according to claim 31, wherein the polymer is selected from the group consisting of polyethylene oxide (PEO), polyethylene glycol (PEG), amino-terminated polyethylene oxide (PEO-NH₂), amino-terminated polyethylene glycol (PEG-NH₂), polypropylene glycol (PPG), polypropylene oxide (PPO), polypropylene glycol bis(2-amino-propyl ether) (PPG-NH₂), polyvinyl alcohol, poly(acrylic acid), poly(vinyl pyridine), poly(N-isopropylacrylamide) (polyNIPAM) and polyallylamine (PAM).

33. (Original) The method according to claim 32, wherein the water soluble polymer is selected from the group consisting of PEO, PEO-NH₂, PEG, PPG-NH₂, polyNIPAM and PAM.

34. (Original) The method according to claim 33, wherein the polymer is PEO.

35. (Original) The method according to claim 25, wherein the linker is selected from the group consisting of C₁₋₂₀alkylene, C₁₋₂₀alkenylene, organic ethers, thioethers, amines, esters, amides, urethanes, carbonates and ureas.

36. (Original) The method according to claim 25, wherein the compound of Formula I is selected from one or more of:

GluconamideSi (Compound 1);

MaltonamideSi (Compound 2);

DextronamideSi (Compound 3);

(CH₂CH₂O)_p[(EtO)₃Si(C₃H₆)]₂, p ~4-5, average MW 200 (Compound 5a);

(CH₂CH₂O)_p[(EtO)₃Si(C₃H₆)]₂, p ~13, average MW 600 (Compound 5b);

(CH₂CH₂O)_p[(EtO)₃Si(C₃H₆)]₂, p ~44, average MW 2000 (Compound 5c); and

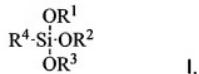
(CH₂CH₂O)_p[(EtO)₃Si(C₃H₆)]₂, p ~227, average MW 10,000 (Compound 5d).

37. (Original) The method according to claim 1, wherein the organic polyol silane precursor is selected from the group consisting of diglycerylsilane (DGS), monosorbitylsilane (MSS), monomaltosylsilane (MMS), dimaltosylsilane (DMS) and dextran-based silane (DS).

38. (Currently Amended) The method according to claim 1, wherein the conditions suitable for the hydrolysis and condensation of the precursor to a siliceous material include a pH in the range of about 4-11.5 comprise combining the organic polyol silane precursor with the one or more additives in aqueous solutions and with optional sonication to assist in dissolution.

39. (Currently amended) A method of preparing siliceous materials with low shrinkage characteristics comprising:

- (a) combining an aqueous solution of one or more compounds of Formula I:

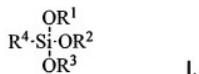


wherein OR¹, OR² and OR³ are the same or different and represent a group that is hydrolyzed under normal sol-gel conditions to provide Si-OH groups; and R⁴ is group that is not hydrolyzed under normal sol-gel conditions, with an aqueous solution of an organic polyol silane precursor;

- (b) adjusting the pH of the solution in (a) to about 4-11.5;
(c) allowing the solution of (b) to gel;
(d) aging the gel of (c); and
(e) drying the aged gel in air.

40. (Original) A siliceous material prepared using the method according to claim 1.

41. (Currently amended) A method of preparing monolithic silica materials comprising combining an organic polyol silane precursor with one or more additives selected from one or more water-soluble polymers and one or more compounds of Formula I:

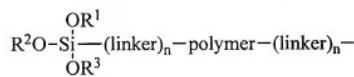


wherein OR¹, OR² and OR³ are the same or different and represent a group that is hydrolyzed under normal sol-gel conditions to provide Si-OH groups, R⁴ is group

selected from polymer-(linker)_n- and $\begin{array}{c} \text{OR}^1 \\ | \\ \text{R}^2\text{O-Si-}(\text{linker})_n-\text{polymer-}(\text{linker})_n- \\ | \\ \text{OR}^3 \end{array}$ and n = 0-1, under conditions where a phase transition occurs before gelation, wherein the conditions where a phase transition occurs before gelation comprise combining the

organic polyol silane precursor with the one or more additives at a pH in the range of about 4 to about 11.5.

42. (Original) The method according to claim 41, wherein R⁴ is



43. (Original) The method according to claim 42, wherein the linker group is a C₁₋₄alkylene group and n is 1.

44. (Original) The method according to claim 42, wherein OR¹, OR² and OR³ are the same and are selected from C₁₋₄alkoxy.

45. (Original) The method according to claim 42, wherein the polymer is PEO.

46. (Original) The method according to claim 41 wherein the compound of Formula I is selected from the group consisting of:

(CH₂CH₂O)_p[(EtO)₃Si(C₃H₆)₂], p ~4-5, average MW 200 (Compound 5a);

(CH₂CH₂O)_p[(EtO)₃Si(C₃H₆)₂], p ~13, average MW 600 (Compound 5b);

(CH₂CH₂O)_p[(EtO)₃Si(C₃H₆)₂], p ~44, average MW 2000 (Compound 5c); and

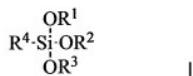
(CH₂CH₂O)_p[(EtO)₃Si(C₃H₆)₂], p ~227, average MW 10,000 (Compound 5d).

47. (Original) The method according to claim 41, wherein the water soluble polymer is selected from one or more of PEO, PEO-NH₂ and poly(NIPAM).

48. (Original) A meso/macroporous silica monolith prepared using the method according to claim 41.

49. (Currently amended) A method of preparing siliceous materials comprising combining an organic polyol silane precursor, a biomolecule of interest and one or more

additives under conditions suitable for the hydrolysis and condensation of the precursor to a siliceous material, wherein the one or more additives are selected from one or more water-soluble polymers and one or more trifunctional silanes of Formula I:



wherein OR¹, OR² and OR³ are the same or different and represent a group that is hydrolyzed under normal sol-gel conditions to provide a Si-OH group; and R⁴ is group that is not hydrolyzed under normal sol-gel conditions, wherein the conditions suitable for hydrolysis and condensation of the precursor to a siliceous material comprise combining the organic polyol silane precursor, biomolecule and one or more additives at a pH in the range of about 4 to about 11.5.

50. (Original) A siliceous material comprising a biomolecule entrapped therein prepared using the method according to claim 49.

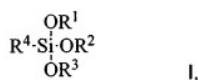
51. (Previously amended) A method for the quantitative or qualitative detection of a test substance that reacts with, binds to and/or whose reactivity is catalyzed by an active biological substance, wherein said biological substance is encapsulated within a siliceous material, comprising:

- (a) preparing the siliceous material comprising said active biological substance entrapped within a porous, silica matrix using a method according to claim 49;
- (b) bringing said biological-substance-containing siliceous material into contact with a gas or aqueous solution comprising the test substance; and
- (c) quantitatively or qualitatively detecting, observing or measuring the change in one or more characteristics in the biological substance entrapped within the siliceous material and/or, alternatively, quantitatively or qualitatively detecting, observing or measuring the change in one or more characteristics in the test substance.

52. (Original) The method according to claim 51, wherein the change in one or more characteristics of the entrapped biological substance is qualitatively or quantitatively measured by spectroscopy, utilizing one or more techniques selected from UV, IR, visible light, fluorescence, luminescence, absorption, emission, excitation and reflection.

53. (Original) A method of storing a biologically active biological substance in a silica matrix, wherein the biological substance is an active protein or active protein fragment, wherein the silica matrix prepared using a method according to claim 49.

54. (Currently amended) A method of preparing a monolithic silica chromatographic column comprising placing a solution comprising an organic polyol silane precursor and one or more additives selected from one or more water-soluble polymers and one or more compounds of Formula I:

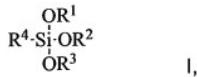


wherein OR¹, OR² and OR³ are the same or different and represent a group that is hydrolyzed under normal sol-gel conditions to provide a Si-OH group; R⁴ is group

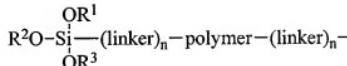
selected from polymer-(linker)_n- and $\begin{array}{c} \text{OR}^1 \\ | \\ \text{R}^2\text{O-Si}-(\text{linker})_n-\text{polymer}-(\text{linker})_n- \\ | \\ \text{OR}^3 \end{array}$ and n = 0-1, in a column under conditions suitable for a phase transition to occur before gelation, wherein the conditions suitable for a phase transition to occur before gelation comprise combining the organic polyol silane precursor with the one or more additives at a pH in the range of about 4 to about 11.5.

55. (Previously amended) The method according to claim 54, wherein the solution further comprises one or more substances, which provide cationic sites that counterbalance an anionic charge of the silica to reduce non-selective interactions

56. (Currently amended) A chromatographic column comprising a silica monolith prepared by combining an organic polyol silane precursor and one or more additives selected from one or more water-soluble polymers and one or more compounds of Formula I:



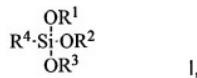
wherein OR¹, OR² and OR³ are the same or different and represent a group that is hydrolyzed under normal sol-gel conditions to provide Si-OH groups; R⁴ is group



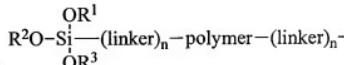
selected from polymer-(linker)_n- and and n = 0-1, under conditions where a phase transition occurs before gelation, wherein the conditions suitable for a phase transition to occur before gelation comprise combining the organic polyol silane precursor with the one or more additives at a pH in the range of about 4 to about 11.5.

57. (Currently amended) A method of preparing a monolithic silica column having an active biomolecule entrapped therein comprising combining:

- a polyol-silane derived silica precursor;
- one or more additives selected from one or more water soluble polymers and one or more compounds of Formula I:



wherein OR¹, OR² and OR³ are the same or different and represent a group that is hydrolyzed under normal sol-gel conditions to provide Si-OH groups, R⁴ is group

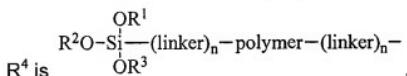


selected from polymer-(linker)_n- and and n is 0-1; and

- a biomolecule;

under conditions wherein a phase separation occurs before gelation, wherein the conditions suitable for a phase transition to occur before gelation comprise combining the organic polyol silane precursor with the one or more additives at a pH in the range of about 4 to about 11.5..

58. (Original) The method according to claim 57, wherein the one or more additives is one or more water soluble polymers or one or more compounds of Formula I, wherein



59. (Previously amended) The method according to claim 57, wherein the organic polyol silane silica precursor, one or more additives and biomolecules are also combined with a substance which provides cationic sites that counterbalance an anionic charge of the silica to reduce non-selective interactions.

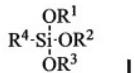
60. (Original) A chromatographic column prepared using a method according to claim 57.

61. (Original) A method of performing immunoaffinity chromatography, sample cleanup, solid phase extraction or preconcentration of analytes, removal of unwanted contaminants, solid phase catalysis or frontal affinity chromatography comprising:

- applying a sample to a column according to claim 60; and
- performing immunoaffinity chromatography, sample cleanup, solid phase extraction or preconcentration of analytes, removal of unwanted contaminants, solid phase catalysis or frontal affinity chromatography.

62. (Previously amended) A method of preparing siliceous materials with enhanced protein stabilizing ability comprising combining an organic polyol silane precursor with one or more additives under conditions suitable for hydrolysis and condensation of

precursor to a siliceous material, wherein the one or more additives is selected from one or more trifunctional silanes of Formula I:



wherein wherein OR¹, OR² and OR³ are the same or different and represent a group that is hydrolyzed under normal sol-gel conditions to provide a Si-OH group and R⁴ is polyol-(linker)-.

63. (Previously amended) The method according to claim 62, wherein the polyol in R⁴ is derived from sugar alcohols, sugar acids, saccharides, oligosaccharides or polysaccharides.

64. (Original) The method according to claim 63, wherein the polyol in R⁴ is derived from allose, altrose, glucose, mannose, gulose, idose, galactose, talose, ribose, arabinose, xylose, lyxose, threose, erythrose, glyceraldehydes, sorbose, fructose, dextrose, levulose, sorbitol, sucrose, maltose, cellobiose, lactose, dextran (500-50,000 MW), amylose, pectin, glycerol, propylene glycol or trimethylene glycol.

65. (Original) The method according to claim 64, wherein the polyol in R⁴ is derived from glycerol, sorbitol, maltose, trehalose, glucose, sucrose, amylose, pectin, lactose, fructose, dextrose or dextran.

66. (Original) The method according to claim 65, wherein the polyol in R⁴ is derived from glycerol, sorbitol, glucose, maltose or dextran.

67. (Original) The method according to claim 66, wherein the polyol in R⁴ is derived from glucose or maltose.

68. (Previously amended) The method according to claim 62 wherein the one or more additives is GluconamideSi (Compound 1) and/or MaltonamideSi (Compound 2).

69. (Original) The method according to claim 62, wherein the protein is a kinase, luciferase, or urease or is Factor Xa.

70. (Original) The method according to claim 69, wherein the protein is Src protein tyrosine kinase.

71. (Original) The method according to claim 62, further comprising combining the organic polyol silane precursor and one or more additives with a substrate for the protein to be entrapped.

72. (Original) The method according to claim 71, wherein the protein is a kinase and the substrate is a source of phosphate.

73. (Original) The method according to claim 72, wherein the substrate is ATP.

74. (Previously added) The method according to claim 59, wherein the substance which provides cationic sites that counterbalance an anionic charge of the silica to reduce non-selective interactions is aminopropyltriethoxysilane (APTES), PAM, PPG-NH₂ and/or PEG-NH₂.